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you for being an important part of keeping this knowledge alive and relevant. Part 1 deals with the theory of misconceptions, by including information on some of the key alternative conceptions that have been uncovered by research. All chemistry students need a basic understanding of quantum theory and its applications in atomic and molecular structure and spectroscopy. This book provides a gentle introduction to the subject with the required background in physics and mathematics kept to a minimum. It develops the basic concepts needed as background. The emphasis throughout is on the physical concepts and their application in chemistry, especially to atoms and to the periodic table of elements Our NEET Foundation series is sharply focused for the NEET aspirants. Most of the students make a career choice in the middle school and, therefore, choose their stream informally in secondary and formally in senior secondary schooling, accordingly. If you have decided to make a career in the medical profession, you need not look any further! Adopt this series for Class 9 and 10 today. This profusely illustrated book, by a world-renowned chemist and award-winning chemistry teacher, provides science students with an introduction to atomic and molecular structure and bonding. (This is a reprint of a book first published by Benjamin/Cummings, 1973.) This book is intended for physicists and chemists who need to understand the theory of atomic and molecular structure and processes, and who wish to apply

the theory to practical problems. As far as practicable, the book provides a self-contained account of the theory of relativistic atomic and molecular structure, based on the accepted formalism of bound-state Quantum Electrodynamics. The author was elected a Fellow of the Royal Society of London in 1992. The #1 New York Times bestseller. Over 4 million copies sold! Tiny Changes, Remarkable Results No matter your goals, Atomic Habits offers a proven framework for improving--every day. James Clear, one of the world's leading experts on habit formation, reveals practical strategies that will teach you exactly how to form good habits, break bad ones, and master the tiny behaviors that lead to remarkable results. If you're having trouble changing your habits, the problem isn't you. The problem is your system. Bad habits repeat themselves again and again not because you don't want to change, but because you have the wrong system for change. You do not rise to the level of your goals. You fall to the level of your systems. Here, you'll get a proven system that can take you to new heights. Clear is known for his ability to distill complex topics into simple behaviors that can be easily applied to daily life and work. Here, he draws on the most proven ideas from biology, psychology, and neuroscience to create an easy-to-understand guide for making good habits inevitable and bad habits impossible. Along the way, readers will be inspired and entertained with true stories from Olympic gold

medalists, award-winning artists, business leaders, life-saving physicians, and star comedians who have used the science of small habits to master their craft and vault to the top of their field. Learn how to: make time for new habits (even when life gets crazy); overcome a lack of motivation and willpower; design your environment to make success easier; get back on track when you fall off course; ...and much more. Atomic Habits will reshape the way you think about progress and success, and give you the tools and strategies you need to transform your habits--whether you are a team looking to win a championship, an organization hoping to redefine an industry, or simply an individual who wishes to quit smoking, lose weight, reduce stress, or achieve any other goal. Originally published in 1938, this book contains ten lectures on subjects such as parasitology, radioactivity, astronomy and evolution theory. Niels Bohr (1885-1962) was a Danish physicist who played a key role in the development of atomic theory and quantum mechanics, he was awarded the Nobel Prize for Physics in 1922. Originally written for various journals during the 1920s, these articles investigate the epistemological significance of discoveries in quantum physics. This book provides a hands-on experience with atomic structure calculations. Material covered includes angular momentum methods, the central field Schrödinger and Dirac equations, Hartree-Fock and Dirac-Hartree-Fock equations, multiplet

structure, hyperfine structure, the isotope shift, dipole and multipole transitions, basic many-body perturbation theory, configuration interaction, and correlation corrections to matrix elements. The book also contains numerical methods for solving the Schrödinger and Dirac eigenvalue problems and the (Dirac)-Hartree-Fock equations. Graduate-level textbook for physicists, chemists and materials scientists. This book is written as a result of a personal conviction of the value of incorporating historical material into the teaching of chemistry, both at school and undergraduate level. Indeed, it is highly desirable that an undergraduate course in chemistry incorporates a separate module on the history of chemistry. This book is therefore aimed at teachers and students of chemistry, and it will also appeal to practising chemists. While the last 25 years has seen the appearance of a large number of specialist scholarly publications on the history of chemistry, there has been little written in the way of an introductory overview of the subject. This book fills that gap. It incorporates some of the results of recent research, and the text is illustrated throughout. Clearly, a book of this length has to be highly selective in its coverage, but it describes the themes and personalities which in the author's opinion have been of greatest importance in the development of the subject. The famous American historian of science, Henry Guerlac, wrote: 'It

is the central business of the historian of science to reconstruct the story of the acquisition of this knowledge and the refinement of its method or methods, and-perhaps above all-to study science as a human activity and learn how it arose, how it developed and expanded, and how it has influenced or been influenced by man's material, intellectual, and even spiritual aspirations' (Guerlac, 1977). This book attempts to describe the development of chemistry in these terms. Learn Atoms which is divided into various sub topics. Each topic has plenty of problems in an adaptive difficulty wise. From basic to advanced level with gradual increment in the level of difficulty. The set of problems on any topic almost covers all varieties of physics problems related to the chapter Atomic Structure. If you are preparing for IIT JEE Mains and Advanced or NEET or CBSE Exams, this Physics eBook will really help you to master this chapter completely in all aspects. It is a Collection of Adaptive Physics Problems in Atoms structure for SAT Physics, AP Physics, 11 Grade Physics, IIT JEE Mains and Advanced , NEET & Olympiad Level Book Series Volume 29 This Physics eBook will cover following Topics for Atoms: 1. Old Atomic Models 2. Rutherford Model 3. Niels Bohr Model 4. State Change & Transition Problems 5. Energy Series 6. Miscellaneous Problems 7. Chapter Test The intention is to create this book to present physics as a most systematic approach to develop a good numerical solving skill. About Author

Satyam Sir has graduated from IIT Kharagpur in Civil Engineering and has been teaching Physics for JEE Mains and Advanced for more than 8 years. He has mentored over ten thousand students and continues mentoring in regular classroom coaching. The students from his class have made into IIT institutions including ranks in top 100. The main goal of this book is to enhance problem solving ability in students. Sir is having hope that you would enjoy this journey of learning physics! In case of query, visit www.physicsfactor.com or WhatsApp to our customer care number +91 7618717227 Concepts of Biology is designed for the single-semester introduction to biology course for non-science majors, which for many students is their only college-level science course. As such, this course represents an important opportunity for students to develop the necessary knowledge, tools, and skills to make informed decisions as they continue with their lives. Rather than being mired down with facts and vocabulary, the typical non-science major student needs information presented in a way that is easy to read and understand. Even more importantly, the content should be meaningful. Students do much better when they understand why biology is relevant to their everyday lives. For these reasons, Concepts of Biology is grounded on an evolutionary basis and includes exciting features that highlight careers in the biological sciences and everyday applications of the concepts at hand.We also strive to show the

interconnectedness of topics within this extremely broad discipline. In order to meet the needs of today's instructors and students, we maintain the overall organization and coverage found in most syllabi for this course. A strength of Concepts of Biology is that instructors can customize the book, adapting it to the approach that works best in their classroom. Concepts of Biology also includes an innovative art program that incorporates critical thinking and clicker questions to help students understand--and apply--key concepts. Until now, popular science has relegated the atom to a supporting role in defining the different chemical elements of the periodic table. This bold new title places its subject center stage, shining the spotlight directly onto the structure and properties of this tiniest amount of anything it is possible to identify. The book covers a huge range of topics, including the development of scientific thinking about the atom, the basic structure of the atom, how the interactions between atoms account for the familiar properties of everyday materials; the power and mystery of the atomic nucleus, and what the mysterious quantum realm of subatomic particles and their interactions can tell us about the very nature of reality. Sparkling text banishes an outdated world of dull chemistry, as it brightly introduces the reader to what everything is made of and how it all works, on the most fundamental level. The aim of this book is to present highly accurate and extensive

theoretical Atomic data and to give a survey of selected calculational methods for atomic physics, used to obtain these data. The book presents the results of calculations of cross sections and probabilities of a broad variety of atomic processes with participation of photons and electrons, namely on photoabsorption, electron scattering and accompanying effects. Included are data for photoabsorption and electron scattering cross-sections and probabilities of vacancy decay formed for a large number of atoms and ions. Attention is also given to photoionization and vacancy decay in endohedrals and to positron-atom scattering. The book is richly illustrated. The methods used are one-electron Hartree-Fock and the technique of Feynman diagrams that permits to include many-electron correlations. This is done in the frames of the Random Phase approximation with exchange and the many-body perturbation theory. Newly obtained and previously collected atomic data are presented. The atomic data are useful for investigating the electronic structure and physical processes in solids and liquids, molecules and clusters, astronomical objects, solar and planet atmospheres and atomic nucleus. Deep understanding of chemical reactions and processes is reached by deep and accurate knowledge of atomic structure and processes with participation of atoms. This book is useful for theorists performing research in different domains of contemporary physics, chemistry and

biology, technologists working on production of new materials and for experimentalists performing research in the field of photon and electron interaction with atoms, molecules, solid bodies and liquids. What Is Time Crystal In condensed matter physics, a time crystal is a quantum system of particles whose lowest-energy state is one in which the particles are in repetitive motion. The system cannot lose energy to the environment and come to rest because it is already in its quantum ground state. Because of this the motion of the particles does not really represent kinetic energy like other motion, it has "motion without energy". Time crystals were first proposed theoretically by Frank Wilczek in 2012 as a time-based analogue to common crystals whereas the atoms in crystals are arranged periodically in space, the atoms in a time crystal are arranged periodically in both space and time. Several different groups have demonstrated matter with stable periodic evolution in systems that are periodically driven. In terms of practical use, time crystals may one day be used as quantum memories. How You Will Benefit (I) Insights, and validations about the following topics: Chapter 1: Time crystal Chapter 2: Time translation symmetry Chapter 3: Crystal structure Chapter 4: Spontaneous symmetry breaking Chapter 5: Condensed matter physics Chapter 6: Quantum mechanics Chapter 7: Zero-point energy (II) Answering the public top questions about time crystal. (III) Real world

examples for the usage of time crystal in many fields. (IV) 17 appendices to explain, briefly, 266 emerging technologies in each industry to have 360-degree full understanding of time crystal' technologies. Who This Book Is For Professionals, undergraduate and graduate students, enthusiasts, hobbyists, and those who want to go beyond basic knowledge or information for any kind of time crystal. Each text in this series provides a concise account of the basic principles underlying a given subject, embodying an independent-learning philosophy and including worked examples. This text covers atomic structure and periodicity. Atomic Physics provides a concise treatment of atomic physics and a basis to prepare for work in other disciplines that are underpinned by atomic physics such as chemistry, biology and several aspects of engineering science. The focus is mainly on atomic structure since this is what is primarily responsible for the physical properties of atoms. After a brief introduction to some basic concepts, the perturbation theory approach follows the hierarchy of interactions starting with the largest. The other interactions of spin, and angular momentum of the outermost electrons with each other, the nucleus and external magnetic fields are treated in order of descending strength. A spectroscopic perspective is generally taken by relating the observations of atomic radiation emitted or absorbed to the internal energy levels involved. X-ray spectra

are then discussed in relation to the energy levels of the innermost electrons. Finally, a brief description is given of some modern, laser based, spectroscopic methods for the high resolution study of the nest details of atomic structure. ALERT: Before you purchase, check with your instructor or review your course syllabus to ensure that you select the correct ISBN. Several versions of Pearson's MyLab & Mastering products exist for each title, including customized versions for individual schools, and registrations are not transferable. In addition, you may need a CourseID, provided by your instructor, to register for and use Pearson's MyLab & Mastering products. Packages Access codes for Pearson's MyLab & Mastering products may not be included when purchasing or renting from companies other than Pearson; check with the seller before completing your purchase. Used or rental books If you rent or purchase a used book with an access code, the access code may have been redeemed previously and you may have to purchase a new access code. Access codes Access codes that are purchased from sellers other than Pearson carry a higher risk of being either the wrong ISBN or a previously redeemed code. Check with the seller prior to purchase. -- Fundamentals of General, Organic, and Biological Chemistry by McMurry, Ballantine, Hoeger, and Peterson provides the background in chemistry and biochemistry essential for allied health students, while ensuring students in other

disciplines gain an appreciation of chemistry's significance in everyday life. Unlike many texts on this subject, it is clear and concise, punctuated with practical and familiar examples from students' personal experiences. An exceptional balance of chemical concepts explains the quantitative aspects of chemistry, and provides deeper insight into theoretical chemical principles. It also sets itself apart by requiring students to master concepts before they can move on to the next chapter. The Seventh Edition focuses on making connections between General, Organic, and Biological Chemistry with a number of new and updated features-including all-new Mastering Reactions boxes, new and updated Chemistry in Action boxes (formerly titled Applications), new and revised chapter problems that strengthen the ties between major concepts in each chapter and practical applications, and much more. 032175011X / 9780321750112 Fundamentals of General, Organic, and Biological Chemistry with MasteringChemistry® Package consists of: 0321750837 / 9780321750839 Fundamentals of General, Organic, and Biological Chemistry 0321776461 / 9780321776464 MasteringChemistry® with Pearson eText -- Access Card -- for Fundamentals of General, Organic, and Biological Chemistry Both the interpretation of atomic spectra and the application of atomic spectroscopy to current problems in astrophysics, laser physics, and thermonuclear plasmas require a thorough knowledge of the Slater-Condon theory

of atomic structure and spectra. This book gathers together aspects of the theory that are widely scattered in the literature and augments them to produce a coherent set of closed-form equations suitable both for computer calculations on cases of arbitrary complexity and for hand calculations for very simple cases. The late Professor Condon and Halis Odabşi collaborate to produce an integrated account of the electron structure of atoms. College Chemistry Quick Study Guide & Workbook: Trivia Questions Bank, Worksheets to Review Homeschool Notes with Answer Key PDF (College Chemistry Study Guide with Answer Key for Self-Teaching/Learning) includes worksheets to solve problems with hundreds of trivia questions. "College Chemistry Study Guide" with answer key PDF covers basic concepts and analytical assessment tests. "College Chemistry Question Bank" PDF book helps to practice workbook questions from exam prep notes. College chemistry quick study guide with answers includes self-learning guide with verbal, quantitative, and analytical past papers quiz questions. College Chemistry trivia questions and answers PDF download, a book to review questions and answers on chapters: atomic structure, basic chemistry, chemical bonding: chemistry, experimental techniques, gases, liquids and solids worksheets for college and university revision notes. College Chemistry workbook PDF download with free sample book covers beginner's questions, textbook's study notes to practice

worksheets. Chemistry quick study guide PDF includes college workbook questions to practice worksheets for exam. "College Chemistry Workbook" PDF, a quick study guide with chapters' notes for NEET/MCAT/GRE/GMAT/SAT/ACT competitive exam. "College Chemistry Worksheets" PDF to review problem solving exam tests from chemistry practical and textbook's chapters as: Chapter 1: Atomic Structure Worksheet Chapter 2: Basic Chemistry Worksheet Chapter 3: Chemical Bonding Worksheet Chapter 4: Experimental Techniques Worksheet Chapter 5: Gases Worksheet Chapter 6: Liquids and Solids Worksheet Solve "Atomic Structure Study Guide" PDF, question bank 1 to review worksheet: Atoms, atomic spectrum, atomic absorption spectrum, atomic emission spectrum, molecules, azimuthal quantum number, Bohr's model, Bohr's atomic model defects, charge to mass ratio of electron, discovery of electron, discovery of neutron, discovery of proton, dual nature of matter, electron charge, electron distribution, electron radius and energy derivation, electron velocity, electronic configuration of elements, energy of revolving electron, fundamental particles, Heisenberg's uncertainty principle, hydrogen spectrum, magnetic quantum number, mass of electron, metallic crystals properties, Moseley law, neutron properties, orbital concept, photons wave number, Planck's quantum theory, properties of cathode rays, properties of positive rays, quantum numbers, quantum theory, Rutherford model

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interpretation of temperature, liquids properties, non-ideal behavior of gases, partial pressure calculations, plasma state, pressure units, solid's properties, states of matter, thermometry scales, and van der Waals equation. Solve "Liquids and Solids Study Guide" PDF, question bank 6 to review worksheet: Liquid crystals, types of solids, classification of solids, comparison in solids, covalent solids, properties of crystalline solids, Avogadro number determination, boiling point, external pressure, boiling points, crystal lattice, crystals and classification, cubic close packing, diamond structure, dipole-dipole forces, dipole induced dipole forces, dynamic equilibrium, energy changes, intermolecular attractions, hexagonal close packing, hydrogen bonding, intermolecular forces, London dispersion forces, metallic crystals properties, metallic solids, metal's structure, molecular solids, phase changes energies, properties of covalent crystals, solid iodine structure, unit cell, and vapor pressure. The Principles of Biology sequence (BI 211, 212 and 213) introduces biology as a scientific discipline for students planning to major in biology and other science disciplines. Laboratories and classroom activities introduce techniques used to study biological processes and provide opportunities for students to develop their ability to conduct research. A knowledge of atomic theory should be an essential part of every physicist's and chemist's toolkit. This

book provides an introduction to the basic ideas that govern our understanding of microscopic matter, and the essential features of atomic structure and spectra are presented in a direct and easily accessible manner. Semi-classical ideas are reviewed and an introduction to the quantum mechanics of one and two electron systems and their interaction with external electromagnetic fields is featured. Multielectron atoms are also introduced, and the key methods for calculating their properties reviewed. Nobel Laureate's lucid treatment of kinetic theory of gases, elementary particles, nuclear atom, wave-corpuscles, atomic structure and spectral lines, much more. Over 40 appendices, bibliography. Connect students in grades 4 and up with science using Learning about Atoms. This 48-page book covers topics such as the development of the theory of the atom, atomic structure, the periodic table, isotopes, and researching famous scientists. Students have the opportunity to create a slide show presentation about elements while using process skills to observe, classify, analyze, debate, design, and report. The book includes vocabulary, crossword puzzles, a quiz show review game, a unit test, and answer keys. Chemistry Quick Study Guide & Workbook: Trivia Questions Bank, Worksheets to Review Homeschool Notes with Answer Key PDF (Chemistry Study Guide with Answer Key for Self-Teaching/Learning) includes worksheets to solve problems with hundreds of trivia questions. "Chemistry Study

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Systems Worksheet Chapter 8: Gases Worksheet Chapter 9: Laboratory Worksheet Chapter 10: Liquids and Solids Worksheet Chapter 11: Mole Concept Worksheet Chapter 12: Oxidation-Reduction Worksheet Chapter 13: Rates of Reactions Worksheet Chapter 14: Solutions Worksheet Chapter 15: Thermochemistry Worksheet Solve "Molecular Structure Study Guide" PDF, question bank 1 to review worksheet: polarity, three-dimensional molecular shapes. Solve "Acids and Bases Study Guide" PDF, question bank 2 to review worksheet: Arrhenius concept, Bronsted-lowry concept, indicators, introduction, Lewis concept, pH, strong and weak acids and bases. Solve "Atomic Structure Study Guide" PDF, question bank 3 to review worksheet: electron configurations, experimental evidence of atomic structure, periodic trends, quantum numbers and energy levels. Solve "Bonding Study Guide" PDF, question bank 4 to review worksheet: ionic bond, covalent bond, dipole-dipole forces, hydrogen bonding, intermolecular forces, London dispersion forces, metallic bond. Solve "Chemical Equations Study Guide" PDF, question bank 5 to review worksheet: balancing of equations, limiting reactants, percent yield. Solve "Descriptive Chemistry Study Guide" PDF, question bank 6 to review worksheet: common elements, compounds of environmental concern, nomenclature of compounds, nomenclature of ions, organic compounds, periodic trends in properties of the elements, reactivity of elements. Solve "Equilibrium Systems Study Guide"

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energy, cooling curves, enthalpy (heat) changes, enthalpy (heat) changes associated with phase changes, entropy, introduction, specific heats. This collection of articles, which were first published in 1958 and written on various occasions between 1932 and 1957, forms a sequel to Danish physician Niels Bohr's earlier essays in Atomic Theory and the Description of Nature (1934). "The theme of the papers is the epistemological lesson which the modern development of atomic physics has given us and its relevance for analysis and synthesis in many fields of human knowledge. "The articles in the previous edition were written at a time when the establishment of the mathematical methods of quantum mechanics had created a firm foundation for the consistent treatment of atomic phenomena, and the conditions for an unambiguous account of experience within this framework were characterized by the notion of complementarity. In the papers collected here, this approach is further developed in logical formulation and given broader application." Niels Bohr and the Quantum Atom is the first book that focuses in detail on the birth and development of Bohr's atomic theory and gives a comprehensive picture of it. At the same time it offers new insight into Bohr's peculiar way of thinking, what Einstein once called his 'unique instinct and tact'. Contrary to most other accounts of the Bohr atom, the book presents it in a broader perspective which includes the reception among

other scientists and the criticism launched against it by scientists of a more conservative inclination. Moreover, it discusses the theory as Bohr originally conceived it, namely, as an ambitious theory covering the structure of atoms as well as molecules. By discussing the theory in its entirety it becomes possible to understand why it developed as it did and thereby to use it as an example of the dynamics of scientific theories. Atomic and Nuclear Chemistry, Volume 1: Atomic Theory and Structure of the Atom presents the modern ideas of the atomic theory and atomic structure against the background of their historical development. Topics covered include the classification of elements; atoms and electrons; the wave mechanical model of the atom; and the determination of atomic weights. This volume is comprised of six chapters and begins by discussing the origin of the atomic theory, focusing on the role of John Dalton, Avogadro's hypothesis, and the introduction to the laws of chemical combination. The chapters that follow look at the work of the early scientists that led to the development of the periodic table of elements; the use of the Avogadro number to determine the actual masses of atoms and molecules; and the structure of the atom. The essential results of the simple wave mechanical treatment are summarized in the next chapter. This book concludes by considering developments in the determination of atomic weights. Some brief notes on the character and personality of

the great scientists who are mentioned throughout the text are included. This book is intended for students and practitioners in the fields of chemistry and physics. The Encyclopedia is a complete and authoritative reference work for this rapidly evolving field. Over 200 international scientists, each experts in their specialties, have written over 330 separate topics on different aspects of geochemistry including geochemical thermodynamics and kinetics, isotope and organic geochemistry, meteorites and cosmochemistry, the carbon cycle and climate, trace elements, geochemistry of high and low temperature processes, and ore deposition, to name just a few. The geochemical behavior of the elements is described as is the state of the art in analytical geochemistry. Each topic incorporates cross-referencing to related articles, and also has its own reference list to lead the reader to the essential articles within the published literature. The entries are arranged alphabetically, for easy access, and the subject and citation indices are comprehensive and extensive. Geochemistry applies chemical techniques and approaches to understanding the Earth and how it works. It touches upon almost every aspect of earth science, ranging from applied topics such as the search for energy and mineral resources, environmental pollution, and climate change to more basic questions such as the Earth's origin and composition, the origin and evolution of life, rock weathering and

metamorphism, and the pattern of ocean and mantle circulation. Geochemistry allows us to assign absolute ages to events in Earth's history, to trace the flow of ocean water both now and in the past, trace sediments into subduction zones and arc volcanoes, and trace petroleum to its source rock and ultimately the environment in which it formed. The earliest of evidence of life is chemical and isotopic traces, not fossils, preserved in rocks. Geochemistry has allowed us to unravel the history of the ice ages and thereby deduce their cause. Geochemistry allows us to determine the swings in Earth's surface temperatures during the ice ages, determine the temperatures and pressures at which rocks have been metamorphosed, and the rates at which ancient magma chambers cooled and crystallized. The field has grown rapidly more sophisticated, in both analytical techniques that can determine elemental concentrations or isotope ratios with exquisite precision and in computational modeling on scales ranging from atomic to planetary. Steve and Susan Zumdahl's texts focus on helping students build critical thinking skills through the process of becoming independent problem-solvers. They help students learn to think like a chemists so they can apply the problem solving process to all aspects of their lives. In CHEMISTRY: AN ATOMS FIRST APPROACH, the Zumdahls use a meaningful approach that begins with the atom and proceeds through the concept of molecules, structure, and bonding, to more

complex materials and their properties. Because this approach differs from what most students have experienced in high school courses, it encourages them to focus on conceptual learning early in the course, rather than relying on memorization and a plug and chug method of problem solving that even the best students can fall back on when confronted with familiar material. The atoms first organization provides an opportunity for students to use the tools of critical thinkers: to ask questions, to apply rules and models and to evaluate outcomes. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version. Building on his widely praised seminars, Brooks explains what current is, how it flows, and how it reacts. He begins by reviewing the nature of current, and then explains current flow in basic circuits, discusses sources that supply and drive current, and addresses the unique problems associated with current on PCBs. Computational Atomic Structure: An MCHF Approach deals with the field of computational atomic structure, specifically with the multiconfiguration Hartree-Fock (MCHF) approach and the manner in which this approach is used in modern physics. Beginning with an introduction to computational algorithms and procedures for atomic physics, the book describes the theory underlying nonrelativistic atomic structure calculations (making use of Brett-Pauli corrections for relativistic effects) and details

how the MCHF atomic structure software package can be used to this end. The book concludes with a treatment of atomic properties, such as energy levels, electron affinities, transition probabilities, specific mass shift, fine structure, hyperfine-structure, and autoionization. This modern, reliable exposition of atomic structure theory proves invaluable to anyone looking to make use of the authors' MCHF atomic structure software package, which is available publicly via the Internet. For beginners and specialists in other fields: the Nobel Laureate's introduction to atomic spectra and their relationship to atomic structures, stressing basics in a physical, rather than mathematical, treatment. 80 illustrations.