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web ionic bond also called electrovalent bond type of linkage formed from the electrostatic attraction between oppositely charged ions in a chemical compound such a bond forms when the valence outermost electrons of one atom web an ion 'aɪ nən 1 is an atom or molecule with a net electrical charge the charge of an electron is considered to be negative by convention and this charge is equal and opposite to the charge of a proton which is considered to be positive by convention web a leading u s general entertainment network ion s positively entertaining lineup features major cable broadcast shows original series and special event programming web ion any atom or group of atoms that bears one or more positive or negative electrical charges positively charged ions are called cations negatively charged ions anions ions are formed by the addition of electrons to or the removal of electrons from neutral atoms or molecules or other ions by combination of ions with other particles or by rupture of a web sep 29 2022 ion definition an ion is an atom or molecule with either a positive or negative electrical charge an ion has a charge because the number of protons does not equal the number of electrons to find the number of protons and electrons check out this

article this inequality comes with the addition or removal of electrons from an atom they web an atom or group of atoms that carries a positive or negative electric charge as a result of having lost or gained one or more electrons 2 a charged subatomic particle such as a free electron ion 2 of 3 abbreviation ionic ion 3 of 3 noun suffix 1 a act or process validation b result of an act or process regulation 2 state or condition web a leading u s general entertainment network ion s positively entertaining lineup features major cable broadcast shows original series and special event programming web jul 20 2022 figure 4 7 2 is a graphical depiction of this process figure 4 7 2 the formation of a chlorine ion on the left the chlorine atom has 17 electrons on the right the chloride ion has 18 electrons and has a 1 charge neutral chlorine atom on left has 17 protons and 17 electrons sodium ion on right has 17 protons and 18 electrons with a web sep 4 2019 an ion is defined as an atom or molecule that has gained or lost one or more of its valence electrons giving it a net positive or negative electrical charge in other words there is an imbalance in the number of protons positively charged particles and electrons negatively charged particles in a chemical species history and meaning web definition of ion noun physics chemistry an electrically charged atom or group of atoms formed by the loss or gain of one or more electrons as a cation positive ion which is created by electron loss and is attracted to the cathode in electrolysis or as an anion negative ion which is created by an electron gain and is attracted to the anode

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